

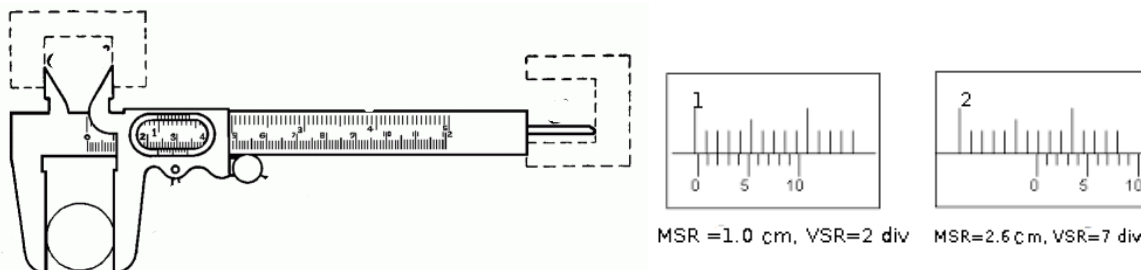
Experiment No:

Date:

Vernier Calipers

- Aim:**
1. To find the volume of a Rectangular Block
 2. To find the internal volume of a Calorimeter
 3. To find mass of water that can be taken in the Calorimeter

Apparatus: Vernier Calipers, Rectangular block, Calorimeter etc



Theory:

Volume of the Rectangular Block = **Length x Breadth x Thickness** of the rectangular block

Inner Volume of the Calorimeter = $\pi r^2 D$

Where r is the inner radius and D is the depth of the Calorimeter given.

All the dimensions are measured using the Vernier Calipers.

Mass of water that can be taken in the calorimeter = Inner Volume of the Calorimeter x Density of Water

Density of Water = **1000 kg/m³**

Total Reading = **MSR + (VSR x LC)**

Where **MSR** is the Main Scale Reading, **VSR** is the Vernier Scale Reading and **LC** is the Least Count of the Vernier Calipers.

$$LC = \frac{1}{n} \times \text{Value of a Main Scale Division}$$

Where **n** is the total number of divisions of the vernier scale

Observations:

Zero Error = Nil

Value of a main Scale division = mm of the Vernier n = div
 1msd = cm $LC = \frac{1}{n} \times 1 \text{msd} = \text{cm}$

Dimension	Sl. No	MSR cm	VSR div	MSR + (VSR x LC) cm	Mean cm
Length of the Rectangular Block	1				L =
	2				
	3				
	4				
	5				
	6				

Dimension	Sl. No	MSR cm	VSR div	MSR + (VSR x LC) cm	Mean cm
Breadth of the Rectangular Block	1				B =
	2				
	3				
	4				
	5				
	6				
Thickness of the Rectangular Block	1				T =
	2				
	3				
	4				
	5				
	6				
Dimension	Sl. No	MSR cm	VSR div	MSR + (VSR x LC) cm	Mean cm
Inner diameter of the Calorimeter	1				d =
	2				
	3				
	4				
	5				
	6				
Depth of the Calorimeter	1				D =
	2				
	3				
	4				
	5				
	6				

Calculations:

$$\text{Volume of the Rectangular Block} = \mathbf{L \times B \times T} = \quad \text{cm}^3$$

$$= \quad \text{m}^3$$

$$\text{Inner diameter of the Calorimeter } d =$$

$$\text{Inner radius of the Calorimeter } r = d/2 =$$

$$\text{Inner Volume of the Calorimeter } \pi r^2 D = \quad \text{cm}^3$$

$$= \quad \text{m}^3$$

$$\text{Mass of water that can be taken in the calorimeter} = \text{Inner Volume of the Calorimeter} \times \text{Density of Water}$$

$$= \quad \text{kg} = \quad \text{kg}$$

Results:

1. Volume of the given Rectangular block = m^3
2. Inner Volume of the given Calorimeter = m^3
3. Mass of water that can be taken in the Calorimeter = kg