Zeroth Law

Zeroth Law of Thermodynamics, which states that 'two systems in thermal equilibrium with a third system separately are in thermal equilibrium with each other' The **Zeroth Law** clearly suggests that when two systems A and B, are in thermal equilibrium, there must be a physical quantity that has the same value for both. This thermodynamic variable whose value is **equal** for two systems in thermal equilibrium is called **temperature** (T)

We have arrived at the concept of temperature formally via the Zeroth Law